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Perovskite and tandem degradation challenges and mitigation strategies

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Content also from Giles Eperon (Swift Solar) & Laura Mundt (Stanford University) – “Perovskite failure modes: What do we know so far?” PVQAT Talk November 2, 2021.

October 6, 2022 Frankfurt am Main, Germany, Degradation modes in new PV cell & module technology

Technology Collaboration Programme

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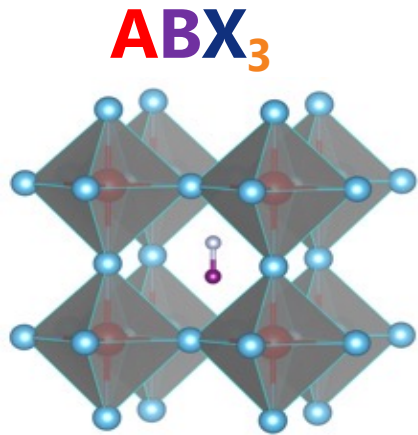
SAND2023-10479C

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Perovskite PV Introduction



- “Perovskite” refers to crystal structure
- Metal Halide Perovskite PV has a range of chemical compositions:

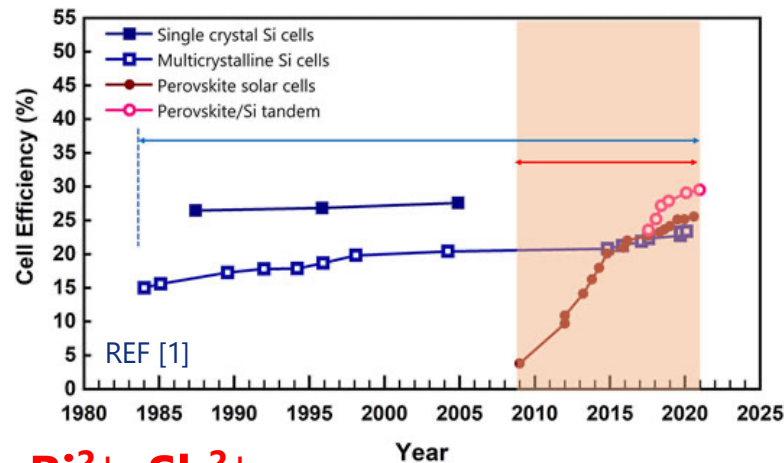


A = Pb²⁺, Sn²⁺, Ge²⁺, Bi²⁺, Sb²⁺, ...

X = Halides (Cl⁻, Br⁻, I⁻)

B = Organics or metal (Cs⁺)

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History

- First perovskite PV cell made in 2009
 - <3% PCE (power conversion efficiency)
- PCE has risen fast
 - $\geq 25.6\%$ today at the cell level (limit – 33%)¹
 - It took 40 years to achieve this PCE for c-Si.
- Promise of low cost manufacturing
 - Low temperature
 - Solution processing
 - High speed manufacturing

¹Zhang et al., 2022: 10.3389/fchem.2022.802890

Degradation Modes for Perovskite PV



Perovskite PV materials are characterized by:

- **Weak bonds** allowing ions to move around in response to electric fields
- Susceptibility to degradation when exposed to **moisture, oxygen, heat & light.**

Packaging needs to **keep water out** and gasses (from heating) inside¹.

- Outgassing reactions are reversible.

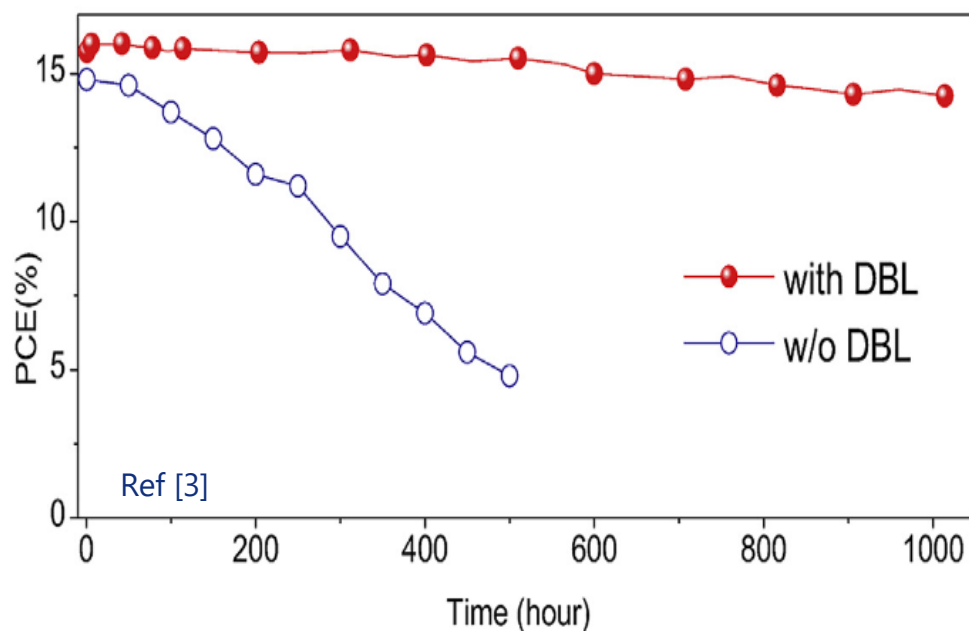
Degradation Mode Classification:

- **Extrinsic** (water, oxygen, mechanical stress, potential induced degradation)
- **Intrinsic** (phase instability, halide segregation, decomposition)
- **Device specific** (electrode diffusion, transport layer reactions, reverse bias)

Extrinsic: Water



Protection against humidity by a diffusion barrier



- Formation of PbI_x ¹
- Can be solved by glass-glass modules², sputtered barriers^{3,4} or back-/frontsheets with diffusion barriers and edge sealing²
- Several reports on modules that withstand damp heat tests for at least 1000 h (IEC 61646)^{2,5}

¹Yang et al., ACS Nano 9, 1955-1936 (2015)

²Cheacharoen et al., Sustainable Energy Fuels, 2, 2398 (2018)

³Bi et al., Joule 3, 2748-2760 (2019)

⁴Ahangharnejhad et al., ACS Appl. Energy Mater., 4, 7571-7578 (2021)

⁵Azmi et al., Science, 376, 6588, 73-77 (2022)

⁶Arias-Ramon et al., Solar Energy Materials and Solar Cells, 215, 110625 (2020)

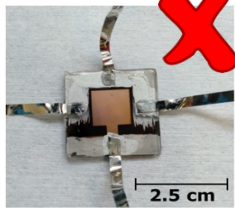
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Mitigation: good encapsulation (diffusion barriers, edge seals, hydrophobic additives)⁶

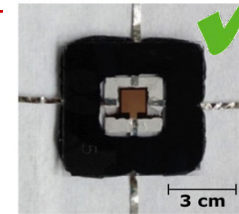
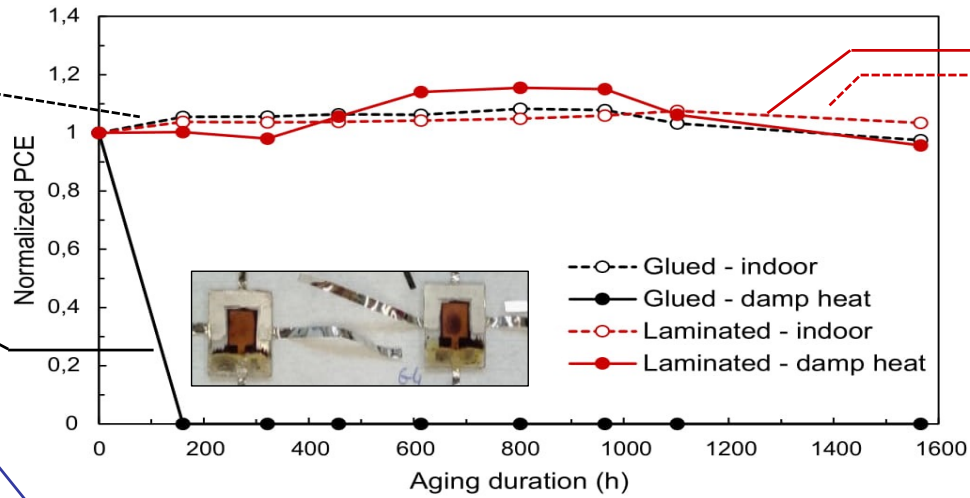
Packaging Examples



DAMP HEAT TEST

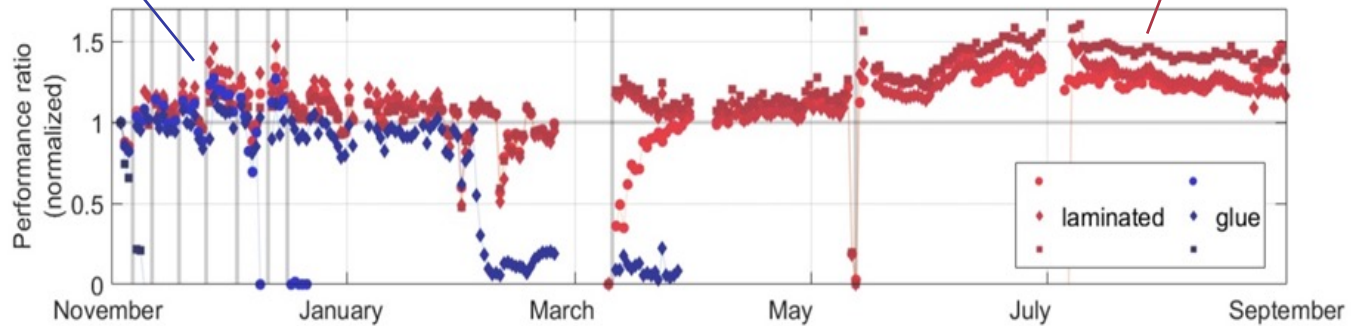


"Glued"



"Laminated"

OUTDOOR

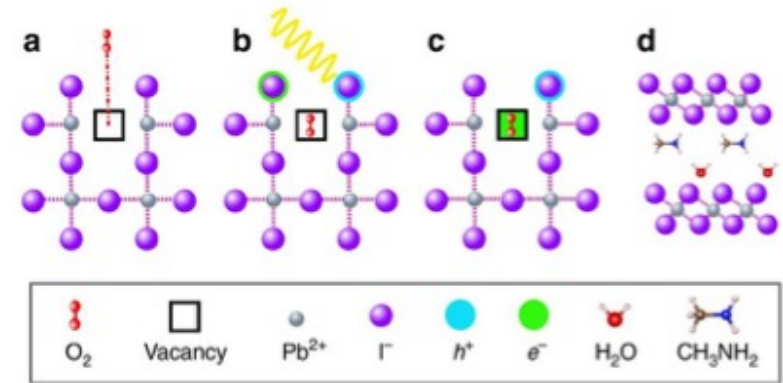


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Extrinsic: Oxygen



- Lead-halide perovskites: oxygen reacts with the perovskite under light.
- MAPI forms superoxides in iodide defect sites.
- Tin perovskites: oxygen oxidises the tin.



Reaction is unfavorable without light



With light, O_2 gathers free electrons and forms a superoxide (O_2^-)



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Mitigation: good encapsulation, cell design with efficient charge extraction, larger perovskite crystals

Aristidou, *Nat. Com.* 2017

Extrinsic: Mechanical Stress



- Mechanical instability of perovskite materials can cause:
 - Issues with delamination
 - Challenges with establishing good electrical contacts
 - Dissimilar thermal expansion coefficients cause stress during thermal cycling
- External stress from wind, hail, snow.

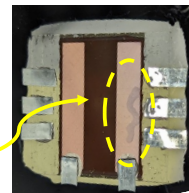
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Mitigation: Matching CTEs for materials, robust packaging, ample testing.

Example of Mechanical Degradation

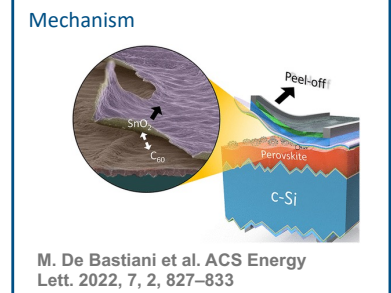
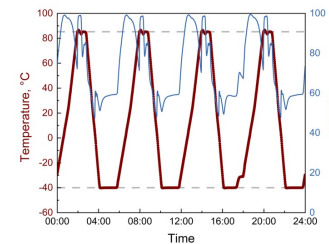
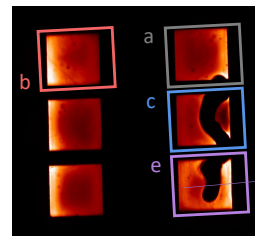
HUMIDITY-FREEZE TEST

Photograph (back side)

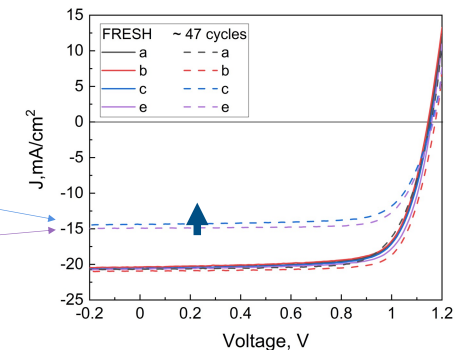


forms after 2-7 cycles

EL images

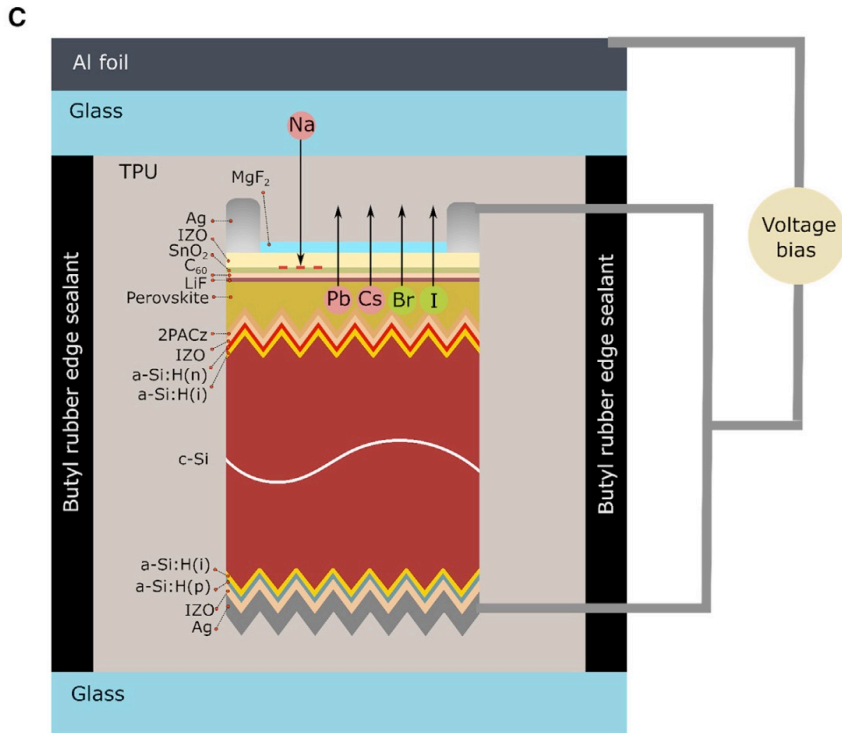


M. De Bastiani et al. ACS Energy Lett. 2022, 7, 2, 827–833



U.Erdil et al., HZB, in preparation

Extrinsic: Potential Induced Degradation



Requires bias + heat

- -1000 V applied for 1 d @ 60 °C to tandem module => 50% PCE loss
- No obvious shunts, Na⁺ ions from glass confined above cells (EDX)
- Diffusion of perovskite elements into encapsulant
- SIMS profile show Cs, Pb, Br and I, as negatively charged ion groups, diffuse along electric field
- Pos. voltage can partially recover PID
- Barriers to prevent diffusion might be needed for commercialization

Xu et al., Cell Reports Physical Science 3, 101026 (2022)

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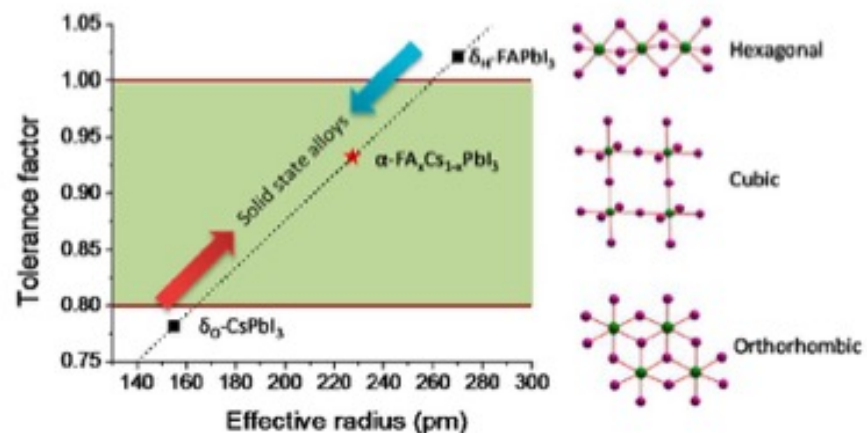
Mitigation: select encapsulant with lower diffusion rates (e.g., POE), other barrier layers

Slide content from Sara Baumann, ISFH

Intrinsic: Phase Instability



- Depending on A-site cation selection or mixture, perovskite can be unstable in photo-active crystal phase at room temperature. Phase change to delta/yellow phase can happen over long timescales in FAPbI_3 .



Li et al., *Chem. Mater.* 2016
Schelhas et al., *EES* 2019

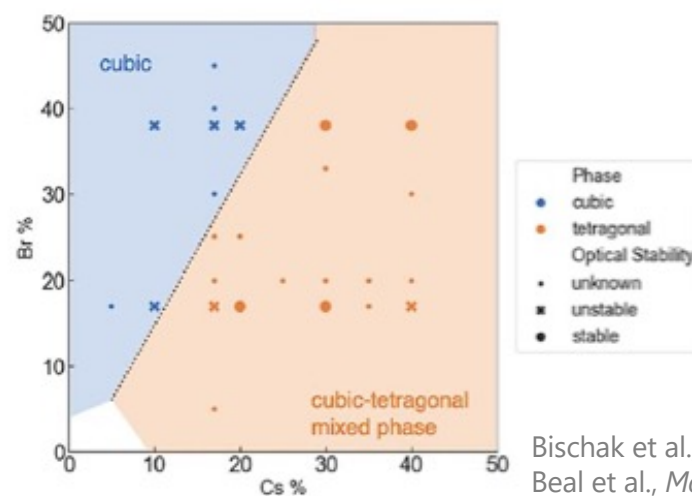
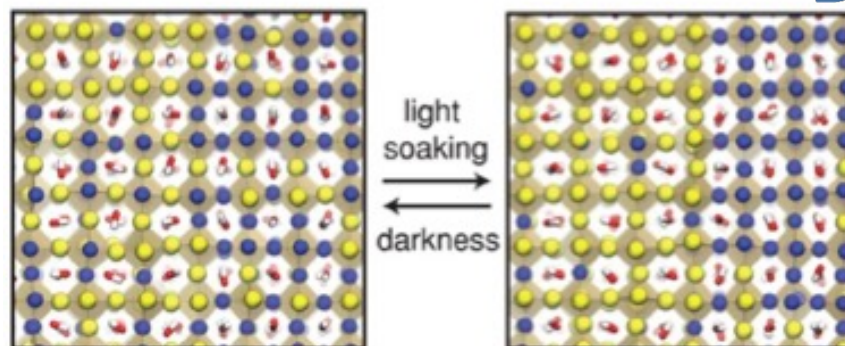
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Mitigation: choose a composition that's in the phase stable regime - note that this space has not been fully explored for all variations of perovskite components yet!

Intrinsic: Halide Segregation



- Under illumination, perovskite compositions with halide ratios (normally I:Br) in a certain regime will undergo halide segregation, limiting voltage and creating defects.



Bischak et al., *Nano Lett.* 2017
Beal et al., *Matter* 2020

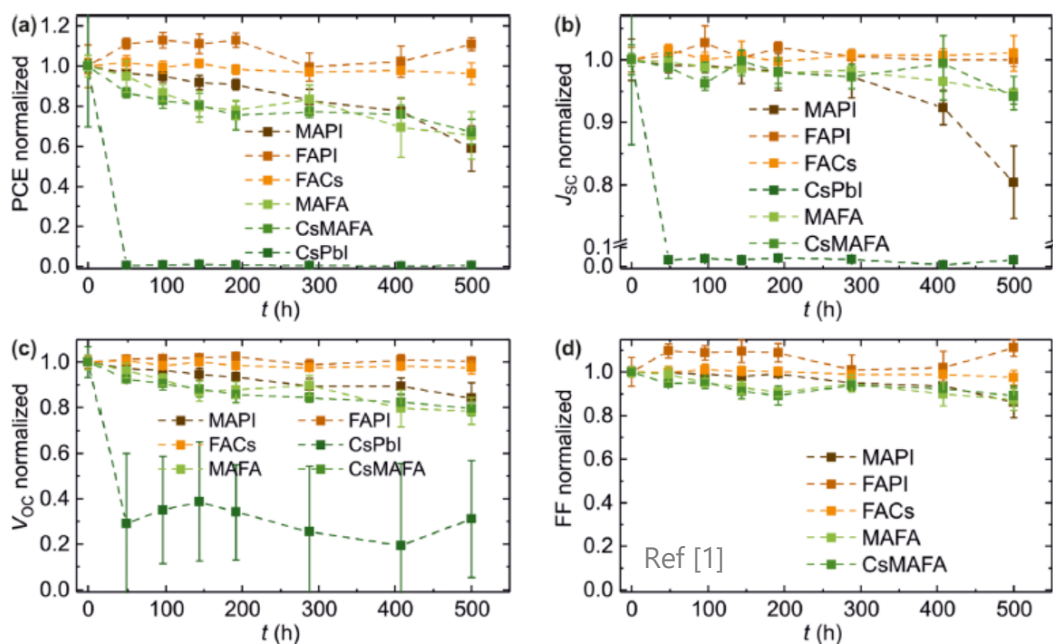
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Mitigation: choose a composition that doesn't segregate but also depends on A-site and B-site.

Intrinsic: Thermal Decomposition



Thermal stability at 85 °C in nitrogen of different perovskite compositions



- Solar cells need to withstand high temperatures during module fabrication (typically > 100 °C) and operation (> 65 °C)²
- MAPbI₃ decomposition (resulting in PbI₂ among others) starts at 85 °C even under air free conditions^{1,3,4}
- Perovskites containing MA less thermally stable than FAPbI₃ and FACsPbI₃¹

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Mitigation: Addition of heat dissipation material (e.g., silicon dioxide particles at perovskite/HTL interface)⁵

¹Schwenzer et al., ACS Appl. Mater. Interfaces, 13, 15292-15304 (2021)

²Raman et al., Renewable and Sustainable Energy Reviews 151, 111608 (2020)

³Conings et al., Adv. Energy Mater., 5, 1500477 (2015)

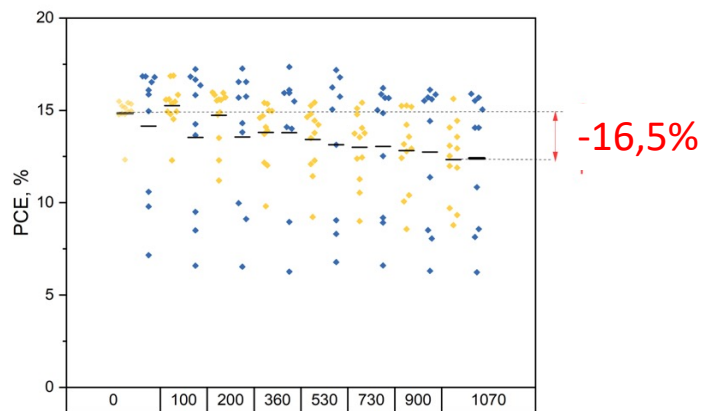
⁴Kim et al., Scientific reports, 7, 4645 (2017)

⁵Pei et al., ACS Energy Lett., 6,9, 3029-3036 (2021)

Thermal Degradation Examples

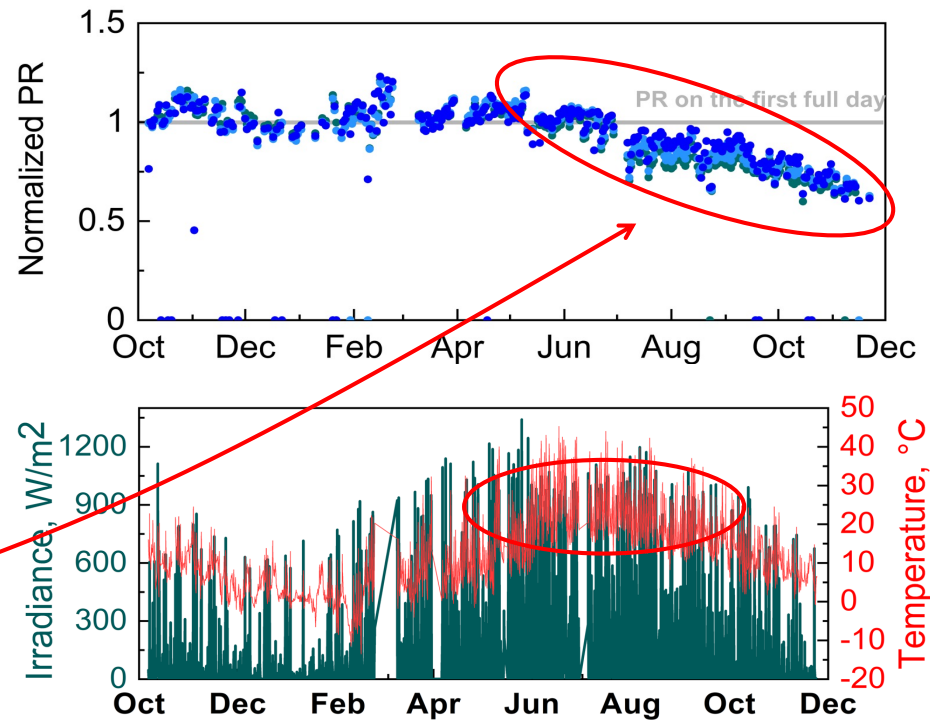


DAMP HEAT TEST



Weak (-ish) thermal stability translates into degradation in summer

OUTDOOR

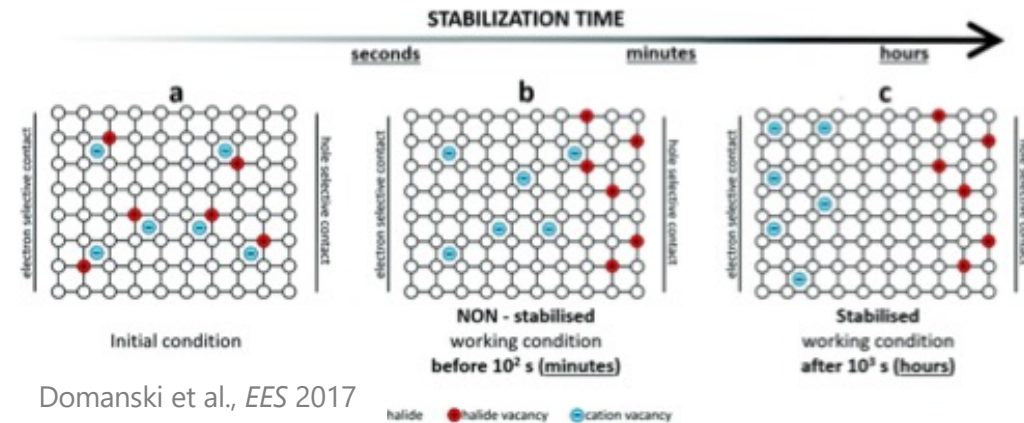


J.Li et al., HZB, submitted

Intrinsic: Ion Migration



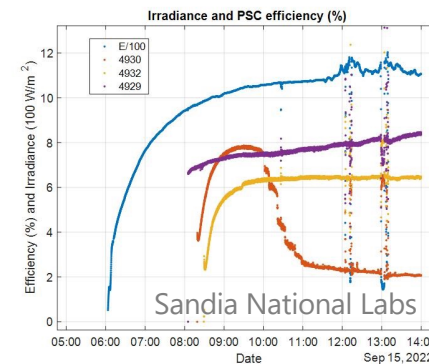
- The constituents of the perovskite lattice (A, X and even B) are able to migrate under low energy inputs
- Mobile ions (most likely iodide) likely shuffle back and forth over day/night cycling.
- Any changes are likely mostly reversible ('morning recovery' has been observed with some stacks), unless there is a reaction between mobile ions at one interface.



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Mitigation: reduce diffusivity

Outdoor MPP data



Recovery?

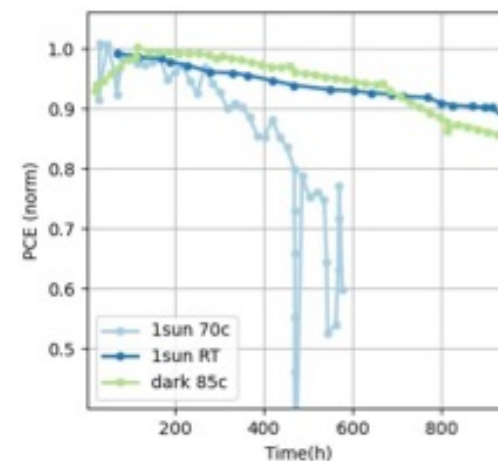
One module fails rapidly

Device: Electrode Diffusion



- Common metals used for top electrode, and some TCOs, diffuse through the stack and react with the perovskite, forming things like AgI, Pb₀ or creating shunt paths.
- If components of the perovskite can diffuse out, they can react with the metal/TCO too.
- Likely more problematic at scribe lines in monolithically interconnected modules.
- This process will happen just under heating, but it is **accelerated dramatically by heat + light conditions** (e.g. operation under elevated temp).

Considered to be one of the most challenging instabilities to make a cost-effective perovskite cell.



Schulz et al., *Chem. Rev.* 2019; Kerner et al., *ACS Energy Lett.* 2020; Schmidt-Mende et al., *APL Materials* 2021

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Mitigation: Use non-metal electrodes (TCOs, carbon), diffusion barrier (e.g., ALD-SNO_x)

Device: Transport Layer Reactions



Various issues with certain HTL or ETL choices:

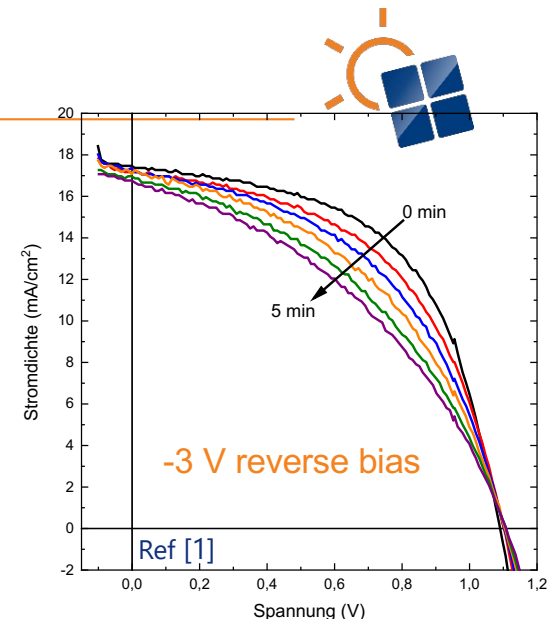
- Reaction with perovskite (e.g. ZnO, NiO, some dopants)
- Thermal instability (e.g. MoO_x, some organics)
- UV or photo-instability (TiO₂)
- Delamination issues

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Mitigation: Selection of appropriate materials for HTL and ETL (e.g. SnO_x, V₂O₅, C₆₀, PTAA, polytpd, etc.)

Device: Reverse Bias

- Partial shading forces cells to operate under reverse bias
- Reverse bias leads to fast and partial irreversible degradation in V_{OC} , I_{SC} , FF & η ¹
- Breakdown differs from avalanche mechanism in cSi cells
- Degradation mechanism based on ion migration^{2,3,4}



Mitigation: (1) Use carbon-based HTL⁵,
 (2) Avoid direct metal contacts⁶,
 (3) Avoid single-cation, single halide perovskites⁶,
 (4) include hole blocking layer between ETL and perovskite^{7,8},
 (5) new interconnection schemes and bypass diodes⁹

¹Richter, master's thesis, ISFH (2021)

²Gould et al., IEEE 48th PVSC, 21129020 (2021)

³Bertoluzzi et al., Adv. Energy Mater., 11, 2002614 (2021)

⁴Bowring et al., DOI: 10.1002/aenm.201702365 (2017)

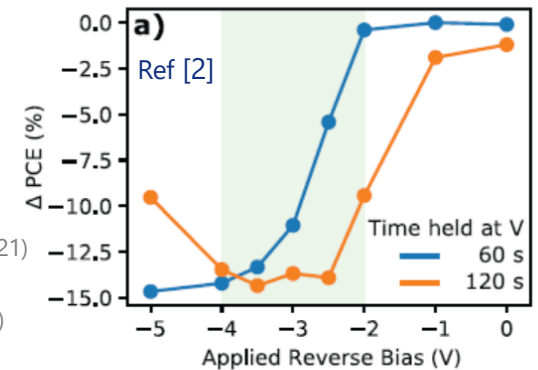
⁵Bowering et al., Adv. Energy Mater. 8, 1702365 (2018)

⁶Bogachuk et al., Sol. RRL, 2100527 (2021)

⁷Gould et al., IEEE 48th PVSC, 21129020 (2021)

⁸Ni et al., Nature Energy, 7, 65-73 (2022)

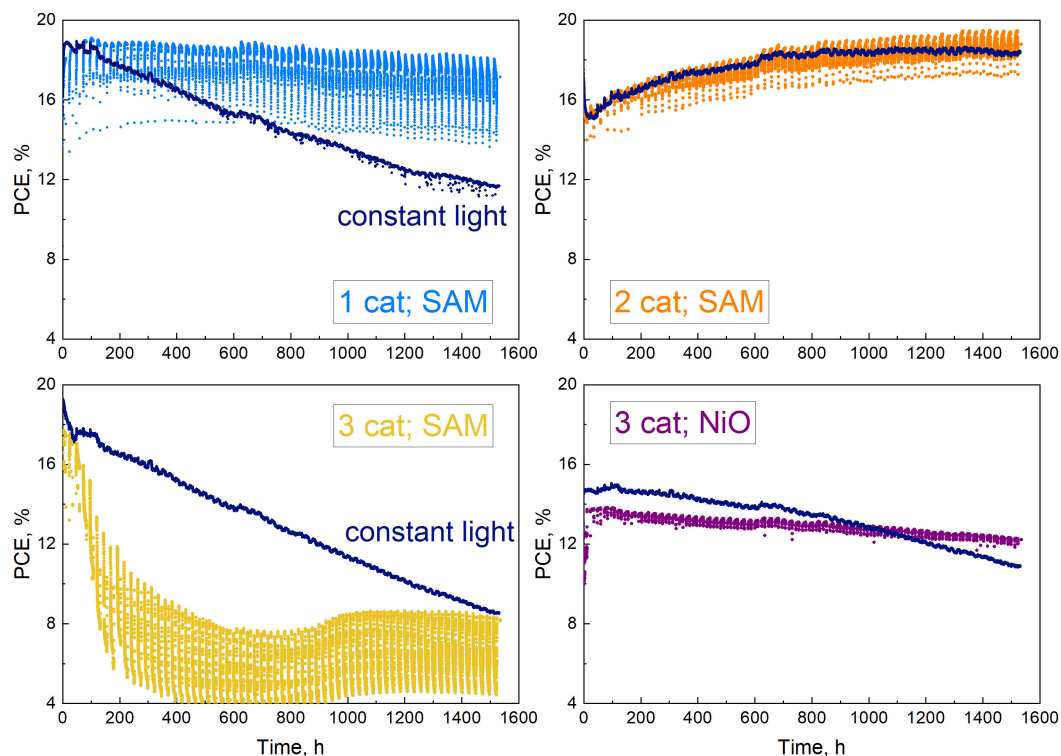
⁹Wolf et al., Sol. RRL, 2, 2100239 (2021)



Metastability



- Affects short-term (within one day/cycle) and long-term stability.
- Not always in a positive way.
- Strongly dependent on the device architecture.
- Translates into the outdoor operation



M. Khenkin, H. Köbler et al., HZB, in preparation

Summary



- Extrinsic, intrinsic and device-specific degradation modes have been summarized.
- Extrinsic modes can be mitigated with effective encapsulation
- Intrinsic modes can be mitigated by selecting the right composition, engineering to increase heat dissipation, and surface engineering.
- Device-specific modes can be mitigated by using non-metal electrodes, diffusion barrier materials, bypass diodes.
- Solutions must be able to scale in size and speed for production.
- More outdoor field tests are needed to identify relevant degradation modes.
- Just because you can induce in a lab does not mean it will be a problem outside (e.g., diurnal and seasonal patterns)

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